organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Mehmet Akkurt,^a* Selvi Karaca,^a Hasan Küçükbay,^b Ersin Orhan^c and Orhan Büyükgüngör^d

^aDepartment of Physics, Faculty of Arts and Sciences, Erciyes University, 38039 Kayseri, Turkey, ^bDepartment of Chemistry, Faculty of Arts and Sciences, Ínönü University, 44280 Malatya, Turkey, ^cDepartment of Chemistry, Faculty of Arts and Sciences, Karaelmas University, 67100 Zonguldak, Turkey, and ^dDepartment of Physics, Faculty of Arts and Sciences, Ondokuz Mayıs University, 55139 Samsun, Turkey

Correspondence e-mail: akkurt@erciyes.edu.tr

Key indicators

Single-crystal X-ray study T = 100 KMean $\sigma(\text{C}-\text{C}) = 0.003 \text{ Å}$ R factor = 0.037 wR factor = 0.100 Data-to-parameter ratio = 16.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

1-Benzyl-3-(2-phenethyl)benzimidazolium bromide monohydrate

The title compound, $C_{22}H_{20}N_2^+\cdot Br^-\cdot H_2O$, was synthesized from 1-benzylbenzimidazole and 2-bromoethylbenzene in dimethylformamide. The dihedral angle between the two phenyl rings is 70.6 (1)°. These phenyl rings make dihedral angles of 52.99 (9) and 83.03 (8)° with the benzimidazole ring system. The crystal structure is stabilized by intermolecular $C-H\cdots Br$ and $O-H\cdots Br$ hydrogen-bond interactions.

Comment

Considerable attention has been given to the synthesis of benzimidazole derivatives because of their therapeutic properties in many diseases. For example, omeprazole, which contains benzimidazole and pyridine, is the best selling antiulcer drug nowadays (Carlsson *et al.*, 2002). The aforementioned compounds show versatile pharmacological activities, such as antibacterial, antifungal, antihelmintic, anti-allergic, antineoplastic, local analgesic, antihistaminic, vasodilative, hypotensive and spasmolytic activities (Easmon *et al.*, 2001; Güneş & Coşar, 1992; Küçükbay *et al.*, 2004). We have also synthesized and investigated the crystal structures of some benzimidazole derivatives (Akkurt *et al.*, 2004, 2005; Türktekin *et al.*, 2004; Karaca *et al.*, 2005). The object of the present study was to elucidate the crystal structure of the title compound, (I).



The molecular structure of (I) is shown in Fig. 1. The bond lengths and angles (Table 1) are comparable with those of similar structures previously reported (Öztürk Yıldırım *et al.*, 2005, Akkurt *et al.*, 2004; Karaca *et al.*, 2005). The benzimidazole ring system is essentially planar, with a maximum deviation of 0.023 (2) Å for atom N1. The two phenyl rings, C9–C14 and C17–C22, make dihedral angles of 83.03 (8) and 52.99 (9)°, respectively, with the benzimidazole ring system.

© 2005 International Union of Crystallography Printed in Great Britain – all rights reserved Received 30 June 2005 Accepted 6 July 2005

Online 9 July 2005



Figure 1

View of (I) with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level.



Figure 2

A view of the crystal packing along the *a* axis. Dashed lines indicate $O-H \cdots Br$ hydrogen bonds.

The dihedral angle between these phenyl rings is 70.6 (1)°. The crystal packing (Fig. 2) is stabilized by intermolecular C– $H \cdots Br$ and O– $H \cdots Br$ hydrogen bonds (Table 2).

Experimental

1-Benzylbenzimidazole (2.0 g, 9.62 mmol) and 2-bromoethylbenzene (1.3 ml, 9.62 mmol) in dimethylformamide (DMF, 3 ml) were heated for 3 h. All the volatiles were then removed under vacuum. The title compound was crystallized from EtOH–Et₂O (3:1 (ν/ν) (yield: 3.28 g, 83%; m.p. 435–436 K). ¹H NMR (DMSO- d_6): δ 3.29 (t, NCH₂CH₂Ph, 2H), 4.84 (t, N–CH₂CH₂Ph, 2H), 5.77 (s, CH₂Ph, 2H), 7.22–8.13 (m, Ar–H, 14H), 9.99 (s, benzimidazole-C²–H, 1H). ¹³C NMR (DMSO- d_6): δ 34.77, 48.50, 50.14, 114.32, 127.12, 127.17, 128.51, 129.06, 129.26, 129.41, 131.06, 131.60, 134.42, 137.27, 142.79. Analysis calculated for

 $C_{22}H_{23}BrN_2O:$ C 64.23, H 5.59, N 6.81%; found: C 63.88, H 5.69, N 6.38%.

Z = 2

 $D_x = 1.422 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation Cell parameters from 10373

reflections $\theta = 2.2-26.8^{\circ}$ $\mu = 2.16 \text{ mm}^{-1}$

Prism colourless

 $0.66 \times 0.47 \times 0.26 \text{ mm}$

3811 reflections with $I > 2\sigma(I)$

 $w = 1/[\sigma^2(F_0^2) + (0.0662P)^2]$

+ 0.4181P] where $P = (F_0^2 + 2F_c^2)/3$

 $\Delta \rho_{\rm max} = 1.45 \text{ e } \text{\AA}^{-3}$

 $\Delta \rho_{\rm min} = -1.06 \text{ e} \text{ Å}^{-3}$

 $(\Delta/\sigma)_{\rm max} = 0.001$

T = 100 K

 $R_{\rm int} = 0.053$

 $\theta_{\rm max} = 26.7^{\circ}$

 $h = -11 \rightarrow 11$

 $k = -11 \rightarrow 11$

 $l = -13 \rightarrow 14$

Crystal data

CarHanN ⁺ ·Br ⁻ ·HaO
$M_r = 410.32$
Triclinic, P1
a = 9.2334 (9) Å
b = 9.3684 (10) Å
c = 11.3028 (12) Å
$\alpha = 99.895 \ (8)^{\circ}$
$\beta = 90.607 \ (8)^{\circ}$
$\nu = 95.637 \ (8)^{\circ}$
$V = 958.14 (17) \text{ Å}^3$

Data collection

Stoe IPDS-II diffractometer ω scans Absorption correction: integration (X-RED32; Stoe & Cie, 2002) $T_{\min} = 0.330, T_{\max} = 0.604$ 10373 measured reflections 4023 independent reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.037$ $wR(F^2) = 0.100$ S = 1.094023 reflections 244 parameters H atoms treated by a mixture of independent and constrained refinement

Table 1

Selected geometric parameters (Å, °).

N1-C1	1.393 (3)	N2-C7	1.334 (3)
N1-C15	1.471 (3)	N2-C8	1.473 (3)
N1-C7	1.331 (3)	N2-C6	1.389 (3)
C1 - N1 - C7	108.76 (17)	N1 - C1 - C2	131.49 (18)
C7-N1-C15	124.55 (17)	N2-C6-C5	131.7 (2)
C1-N1-C15	126.54 (17)	N2-C6-C1	106.60 (16)
C6-N2-C8	126.43 (16)	N1-C7-N2	109.83 (17)
C7-N2-C8	124.91 (17)	N2-C8-C9	111.57 (16)
C6-N2-C7	108.63 (17)	N1-C15-C16	111.07 (16)
N1-C1-C6	106.16 (17)		

Table 2			
Hydrogen-bond geometry	(Å,	°).	

$D - H \cdot \cdot \cdot A$	<i>D</i> -H	Н⋯А	$D \cdots A$	$D - \mathbf{H} \cdots A$
$01-H1W\cdots Br1$ $01-H2W\cdots Br1^{i}$ $C2-H2\cdots Br1$ $C15-H15A\cdots O1^{ii}$	0.84 (4)	2.54 (4)	3.373 (2)	170 (4)
	0.76 (5)	2.61 (5)	3.356 (2)	168 (5)
	0.93	2.82	3.7431 (19)	175
	0.97	2.58	3.458 (3)	151

Symmetry codes: (i) -x, -y, -z + 2; (ii) -x, -y + 1, -z + 2.

The water molecule H atoms were found in difference Fourier maps and refined freely. The other H atoms were positioned geometrically and refined using a riding model, with C–H = 0.93–0.97 Å, and with $U_{\rm iso} = 1.2U_{\rm eq}$ (parent atom). The highest residual peak and the deepest hole are located 0.89 and 0.90 Å, respectively, from atom Br1.

Data collection: X-AREA (Stoe & Cie, 2002); cell refinement: X-AREA; data reduction: X-RED32 (Stoe & Cie, 2002); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

The authors acknowledge the Faculty of Arts and Sciences, Ondokuz Mayıs University, Turkey, for the use of the Stoe IPDS-II diffractometer (purchased under grant F.279 of the University Research Fund). HK and EO thank Ínönü University Scientific Research Unit (BAPB-2005/36) for financial support for this study.

References

Akkurt, M., Karaca, S., Küçükbay, H., Orhan, E. & Büyükgüngör, O. (2005). Acta Cryst. E61, m41-m43.

- Akkurt, M., Öztürk, S., Küçükbay, H., Orhan, E. & Büyükgüngör, O. (2004). Acta Cryst. E60, o219–o221.
- Carlsson, E., Lindberg, P. & Unge, S. (2002). Chem. Br. 5, 42-45.
- Easmon, J., Puerstinger, G., Roth, T., Fiebig, H. H., Jenny, M., Jaeger, W., Heinisch, G. & Hofmann, J. (2001). Int. J. Cancer, 94, 89–96.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838. Güneş, H. S. & Coşar, G. (1992). Arzneim. Forsch. (Drug Res.), 42, 1045-
- 1048. Karaca, S., Akkurt, M., Yılmaz, U., Küçükbay, H. & Büyükgüngör, O. (2005). *Acta Cryst.* E**61**, o2128–o2130.
- Küçükbay, H., Durmaz, R., Okuyucu, N., Günal, S. & Kazaz, C. (2004). Arzneim. Forsch. (Drug Res.), 54, 64–68.
- Öztürk Yıldırım, S., Akkurt, M., Küçükbay, H., Orhan, E. & Büyükgüngör, O. (2005). Acta Cryst. E**61**, o2038–o2039.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Stoe & Cie (2002). *X-AREA* (Version 1.18) and *X-RED32* (Version 1.04). Stoe & Cie, Darmstadt, Germany.
- Türktekin, S., Akkurt, M., Şireci, N., Küçükbay, H. & Büyükgüngör, O. (2004). *Acta Cryst.* E60, 0817–0819.